Announcements

- Turn on the Clicker (the red LED comes on).
- Push "Join" button followed by "20" followed by the "Send" button (switches to flashing green LED if successful).

- Flatulence lab handout is available in the lab handout section of the class web site.
- First Exam 1 week from today. (Sample exam on Web Site).
- Download and read the lab handout for the flatulence lab before your lab next week.
- Free tutors are available through CAR and I have some names for people willing to do private tutoring for a fee. See the Tutors section of the web site.
- Still not signed up for E-mail list: C. Larson, A. Robson-Domin, K. Vorpahl.

Review

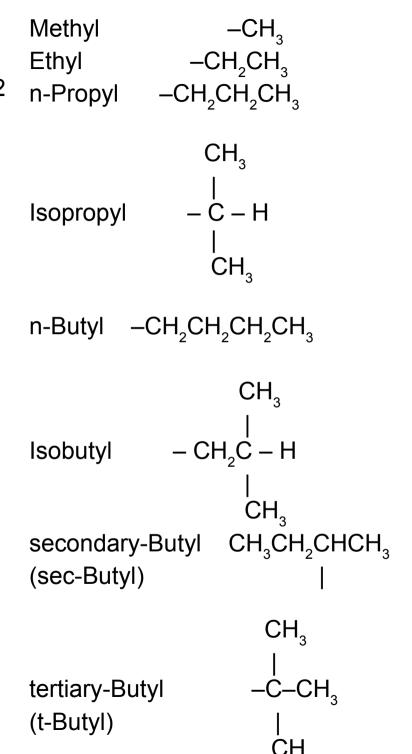
- Petroleum is separated (refined) by fractional distillation.
 - Raoult's law $P_{tot} = X_1 P_1^0 + X_2 P_2^0 + ...$ used to show how distillation works.
 - Branched alkanes are harder to ignite (higher octane)
 - cycloalkanes (other than 5 membered are puckered to maintain 109° sp³ bond angles)
 - aromatic compounds (resonance = delocalization of π electrons = greater stability)
- Oxygenates also added to gasoline (alcohols and ethers).

Naming Branched

Alkanes

Chang table 11.2

- 1)Find the longest chain (backbone).
- 2)Find the largest side group and number the backbone from the end nearest this side chain.
- 3)Use the numbers to indicate the positions of the side groups.
- 4)Alphabetize the side groups.



Alcohols and Ethers

- Alcohols (C-O-H)
 - Name by replacing

 ane of the
 corresponding alkane
 with -ol.
- Examples:
 - CH₃OH = methanol
 - CH₃CH₂CH₂OH = propanol

- Ethers (C-O-C)
 - Name by placing the names of the two alkyl groups in alphabetical order before the word ether.
 - Ignore the prefixes tert-, iso- and sec-.
- Examples
 - tert-butyl methyl ether
 - (CH₃)₃COCH₃

Starch and Cellulose

See carbohydrate handout for cartoon