

Announcements

To join clicker to class today (Clickers with LCD display joins automatically):

- Turn on the Clicker (the red LED comes on).
- Push “Join” button followed by “20” followed by the “Send” button (switches to flashing green LED if successful).

- Chem Club should be selling goggles in the common area between lecture pits after class today.

" Lab next week is *Line Spectra and Significant Digits*. **THERE IS A PRELAB YOU MUST HAVE COMPLETED WHEN YOU COME TO LAB NEXT WEEK.** Watch for the lab handout on the e-mail discussion list or download it from the class web site.

" Exam 1 a week from today.

Samples of exams 1 & 2 from last year are available on class web site in the Study-Aids section.

All other review material (answer keys and notes on end of chapter review material) will be posted in next day or so.

Review

- " Overview of periodic trends.
- " Mass spectrometry. (Charged particles follow different curved paths in a magnetic field depending upon m/z)
- " Isotopes and average atomic mass.

Example H + D:

$$1.007825(0.99985) + 2.014108(0.00015) = 1.0080 \text{ amu}$$
$$= m_{\text{H}}f_{\text{H}} + m_{\text{D}}f_{\text{D}} = m_{\text{avg}}$$

- " Writing nuclear reactions. (sum of mass #'s on lhs = sum of mass #'s on rhs, same for sum of charges)
- " Fusion in stars converting the H produced after the Big Bang into heavier elements.

Chang Fig. 21.2

Cartoons of Radioactive Decay

1. Alpha decay: ${}^m_p X \rightarrow {}^4_2\alpha(\text{expelled}) + {}^{m-4}_{p-2} Y$
2. β decay: $n(\text{in nucl}) \rightarrow p(\text{in nucl}) + {}^0_{-1}e(\text{expelled})$
3. positron emission, $p(\text{in nucl}) \rightarrow n(\text{in nucl}) + {}^0_1e(\text{expelled})$
4. electron capture, $p(\text{in nucl}) + {}^0_{-1}e(\text{falls in}) \rightarrow n(\text{in nucleus})$

Band of Stability

- " Isotopes above the band of stable isotopes (green) undergo β decay to decrease n:p ratio.
- isotopes below band of stable isotopes undergo electron capture or positron emission to increase n:p
- Those out beyond the band of stable isotopes (heavier than $^{209}_{83}\text{Bi}$) are all radioactive and also undergo α decay to lose both n and p

Chang 21.1

Geiger Counter

Chang Fig. 21.18