

# Announcements

To join clicker to class today  
(Clickers with LCD display  
join automatically):

- Turn on the Clicker (the red LED comes on).
- Push “Join” button followed by “20” followed by the “Send” button (switches to flashing green LED if successful).

- Quiz today in discussion.
- Quiz covers everything since exam through VSEPR.
- Will start next section: behavior of gases Friday.

- Suggested reading and problems have been e-mailed out and posted.
- **No shorts, sandals or skirts allowed in Lab!!**
- Volunteer to help with Earth Day Science Fun

# Review

- VSEPR shapes determined by the number of electron groups around the central atom. **YOU MUST KNOW NAMES FOR EACH OF THE SUB-SHAPES.**
- Valence bond theory describes bonding in terms of overlap of hybridized atomic orbitals. Example: sp-hybrid.

